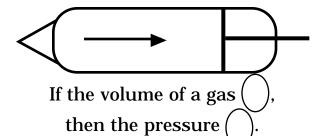
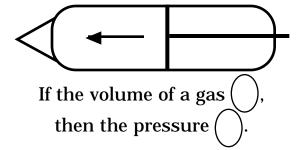
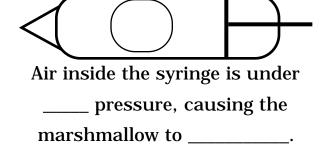
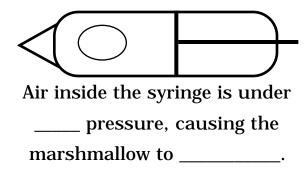
Boyle's Law

states that as the volume of a gas changes, so does its pressure.



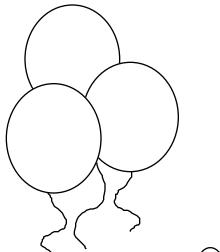


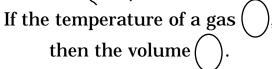


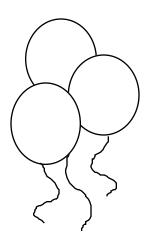


Charles' Law

states that as the temperature of a gas changes, so does its volume.



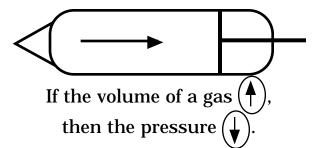


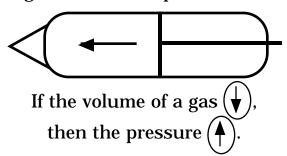


If the temperature of a gas (), then the volume ().

Boyle's Law

states that as the volume of a gas changes, so does its pressure.







Air inside the syringe is under LOW pressure, causing the marshmallow to EXPAND.

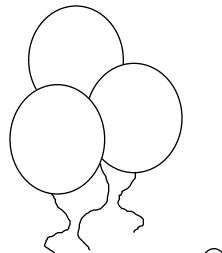


Air inside the syringe is under

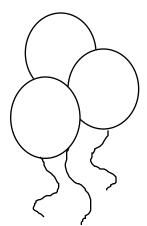
https://example.com/html/>
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Charles' Law

states that as the temperature of a gas changes, so does its volume.



If the temperature of a gas () then the volume ().



If the temperature of a gas $(\begin{tabular}{c} \begin{tabular}{c} \$